

17 Gauss Way Berkeley, CA 94720-5070 p: 510.642.0143 f: 510.642.8609 www.msri.org

#### NOTETAKER CHECKLIST FORM

(Complete one for each talk.)

Email/Phone: mmarciniak@lagcc.cuny.edu 5734620411 Name: Malgorzata Marciniak

Speaker's Name: Kristin Persson

Talk Title: The Materials Project – A Google of Materials

Date: 10 / 04 /2018 Time: <u>2</u>:<u>00</u> am / **pm** (circle one)

Please summarize the lecture in 5 or fewer sentences:

Modern innovation challenges require fast methods to commercialize materials. In the past it took over 18 years. The Materials Project (www.materialsproject.org) is harnessing the power of supercomputing and quantum mechanical theory to compute the properties of known inorganic materials, design novel materials, and offer the data to public. The current release contains data derived from quantum mechanical calculations for over 80,000 materials and millions of associated properties.

#### CHECK LIST

(This is NOT optional, we will not pay for incomplete forms)

Introduce yourself to the speaker prior to the talk. Tell them that you will be the note taker, and that you will need to make copies of their notes and materials, if any.

Obtain ALL presentation materials from speaker. This can be done before the talk is to begin or after the talk; please make arrangements with the speaker as to when you can do this. You may scan and send materials as a .pdf to yourself using the scanner on the 3<sup>rd</sup> floor.

- **Computer Presentations:** Obtain a copy of their presentation •
- Overhead: Obtain a copy or use the originals and scan them •
- Blackboard: Take blackboard notes in black or blue PEN. We will NOT accept notes in pencil • or in colored ink other than black or blue.
- Handouts: Obtain copies of and scan all handouts

For each talk, all materials must be saved in a single .pdf and named according to the naming convention on the "Materials Received" check list. To do this, compile all materials for a specific talk into one stack with this completed sheet on top and insert face up into the tray on the top of the scanner. Proceed to scan and email the file to yourself. Do this for the materials from each talk.

↓ When you have emailed all files to yourself, please save and re-name each file according to the naming convention listed below the talk title on the "Materials Received" check list. (YYYY.MM.DD.TIME.SpeakerLastName)



Email the re-named files to notes@msri.org with the workshop name and your name in the subject line.

# The Materials Project for Accelerated Materials Design and Discovery

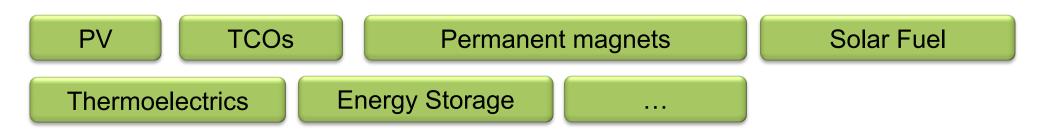
## Kristin A. Persson

UC Berkeley/LBNL

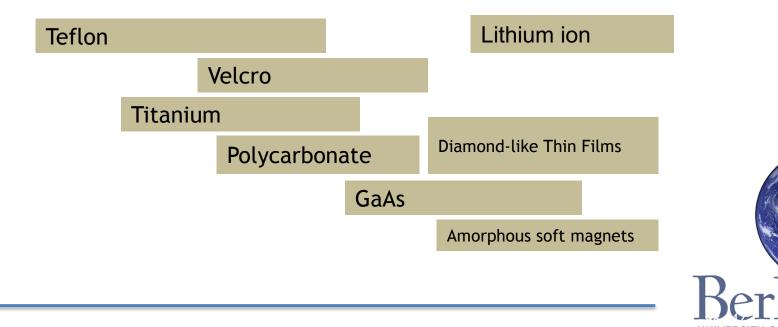


Supported by Department of Energy: Basic Energy Sciences

## **Materials Innovation Timeline**



# How to accelerate the innovation and development timeline?



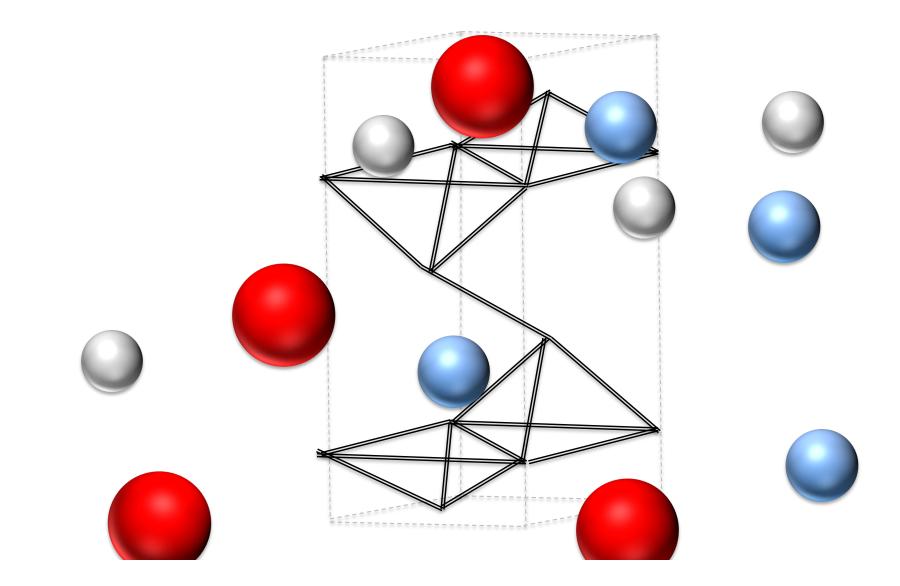
## We do not have materials data

There are about **50,000 to 100,000** known inorganic compounds

Elastic constants: about 200 compounds Super conductors  $\approx 1000$ Dielectric constant  $\approx 300-400$ 

For almost every property we are below 1% in coverage ....





## **Quantum Mechanics**

 $\langle \Psi | \text{better battery} | \Psi^* \rangle$ 



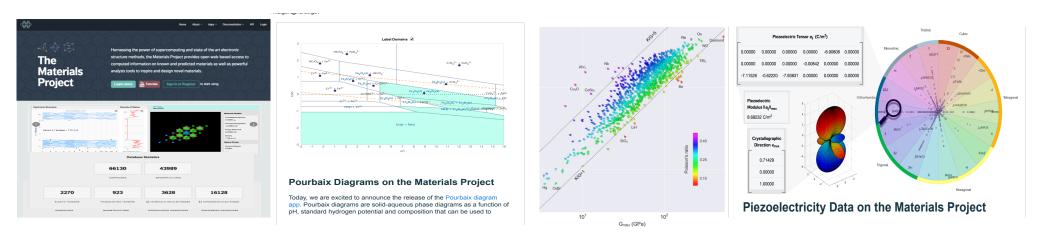
### **Better materials**



## First-Principles Modeling: 1996 - 2018

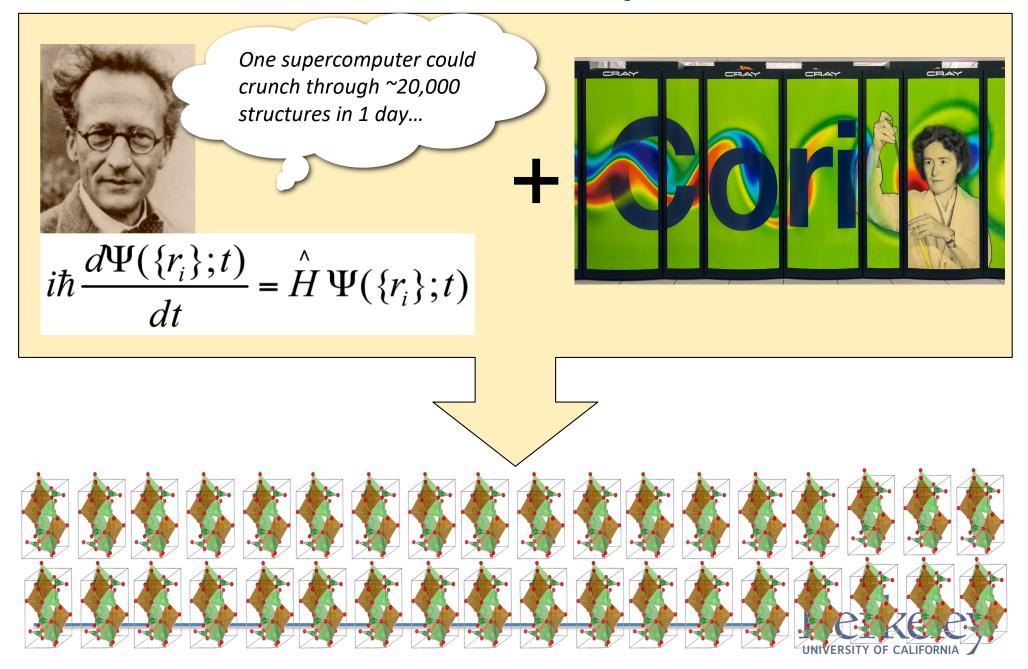
1996: One year on tungsten  $\left( \begin{array}{c} 1000 \\ 0.5 \\ 0.5 \\ 0.5 \\ 0.5 \\ 0.5 \\ 0.0$ 

#### 2018: 80,000+ materials on the Materials Project + millions of materials properties





## **Accelerated Computations**

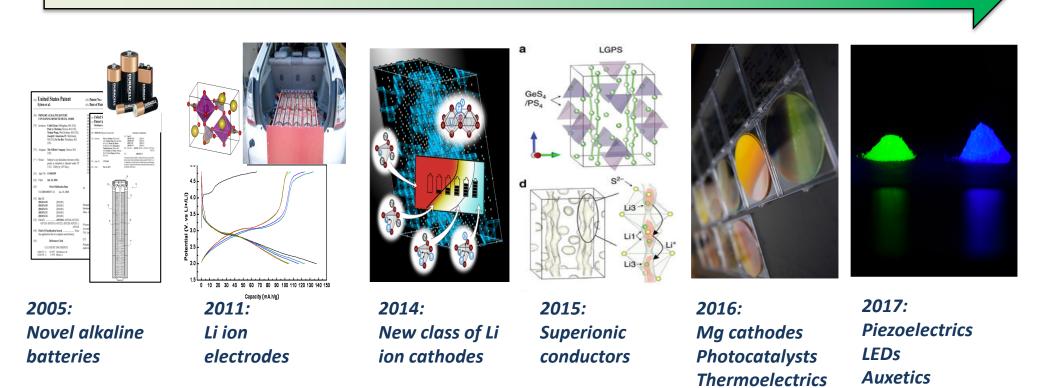


Edisonian Ì  $\square$ In Silico 



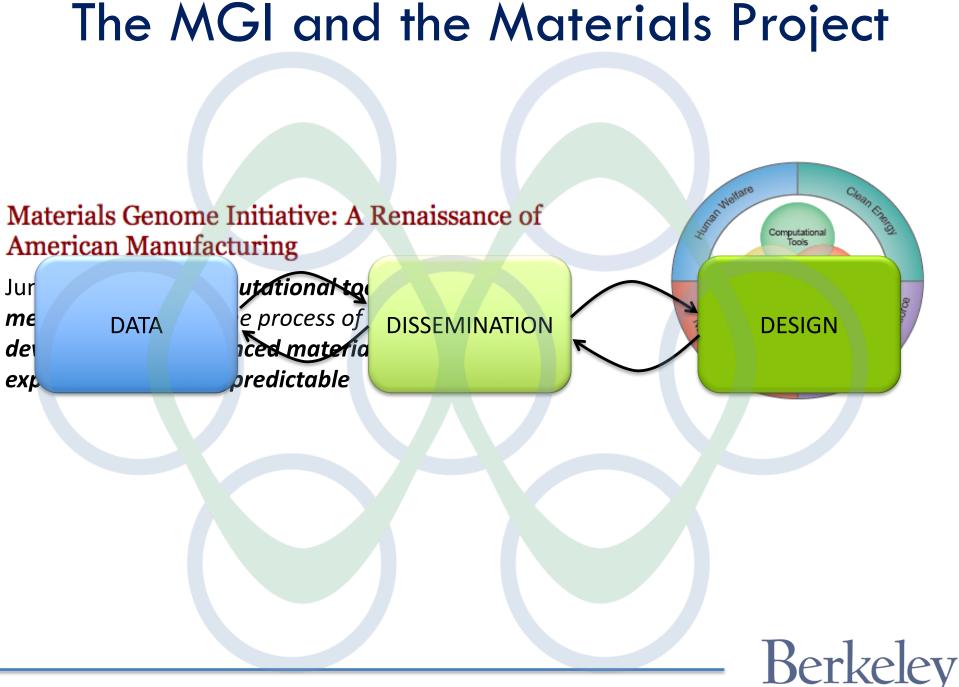
## New Materials for Sustainable Energy

Completely new materials predicted and synthesized based on computational predictions...it is just the beginning





Mg electrolytes



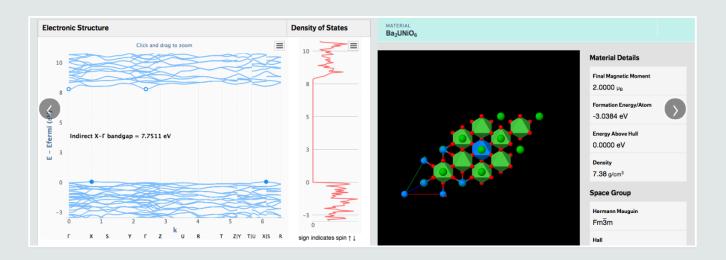
## Materials Project web site

### The **Materials Project**

Harnessing the power of supercomputing and state-of-the-art electronic structure methods, the Materials Project provides open web-based access to computed information on known and predicted materials as well as powerful analysis tools to inspire and design novel materials.

Learn more

to start using Sign In or Register



EXPLORE MATERIALS Search for materials information by chemistry, composition, or property **EXPLORE BATTERIES** Find candidate materials for lithium batteries. Get voltage profiles and oxygen evolution data.

VISUALIZE STABILITY

Generate phase and pourbaix diagrams to find stable phases and study reaction pathways

INVENT **STRUCTURES** Design new compounds

with our structure editor

and substitution

algorithms

#### CALCULATE Calculate the enthalpy of 10,000+ reactions and

compare with experimental values





he

**Materials Explorer Battery Explorer Crystal Toolkit** Structure Predictor **Phase Diagram Pourbaix Diagram Reaction Calculator** 

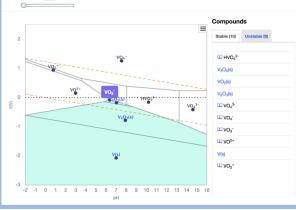
- Nanoporous Explorer
- **Molecules Explorer**

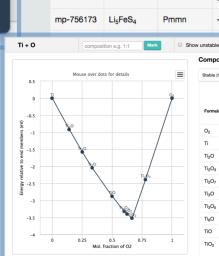
**RFB** Dashboard in mol/ka

Concentration

V 1e-8 0

1





Formula

4

3

19

37

55

10

Materials Id 🔶

**Q** \* ເນ Explore Materials # of elements Q by Elements Fe Li S search × н He 4 9 10 6 Li Be в С 0 F Ne Cl Br 11 12 13 14 15 17 18 16 CI Na Ma Si Ρ S AI Ar Submit 20 24 27 28 29 30 31 32 33 34 35 36 Sc Cu Zn Ga Br κ Ca Ti Cr Mn Fe Co Ni Ge As Se Kr Material Tags 43 44 50 51 54 41 42 45 47 52 53 38 46 48 49 Pd Ag Rb Sr Tc Ru Rh Cd In Sn Sb Те Xe Y Nb Mo imgreite Zr 56 75 76 77 79 80 81 83 72 73 74 78 82 TI Pb Bi At Rn Ba W Re Os Pt Au Hg Po Band Gap (eV) Cs La-Lu Hf Та lr. 89-103 104 105 106 107 108 109 Ra Ac-Lr Rf Db Sg Bh Hs Mt Ds Rg Cn Fr.

57 La		<sup>60</sup> Nd						
		92 U						

#### Nelements 8 Elements records per page ✓ Batch Structures Edit Structures

Formation E Above Hull Band Gap Density Nsites Volume Spacegroup Energy (eV) (eV) (eV) (gm/cc) -1.069 0.062 0.014 20 1.99 365.138 0.022 80 1.991 1459.637 Show labels Compounds 0 25 1.944 96 Stable (10) Unstable (85 0.022 22 2.023 0.022 80 2.076 COL Decomp 540.747 0.021 30 1.866 mp-12957 mp-72 ٥ 3.652 57.712 0 4 -2.043 mp-1215 -3.318 mp-458 Ti<sub>2</sub>O<sub>3</sub> 0 22 2.106 356.072 Ti<sub>2</sub>O<sub>7</sub> -2.39 mp-656850 0.263 10 2.346 189.512 Ti<sub>3</sub>O -1.575 mp-2591 Ti<sub>3</sub>O<sub>5</sub> -3.398 mp-1147 0.008 28 2.422 474.135 mp-554098 -0.918 -2.877 mp-1203 -3.518 mp-554278 ← Previous 1 2 3 Next

Show / hide columns

e.g., 4 or >2 & <6 excluded elements 10 Energy Above Hull Ō 6 Formation Energy # unit cell sites 296 Density 24.6 Volume 7697. **Crystal Systems** Any – Spacegroup Number Any – Spacegroup Symbol Any -Has bandstructure

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Print Exp

Copy

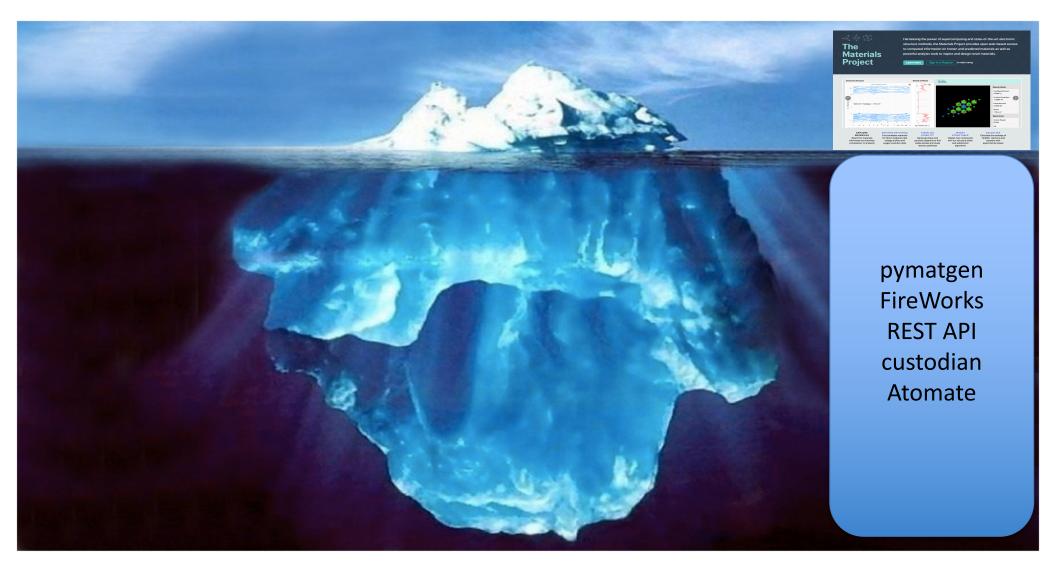
#### Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property Image: Search for materials information by chemistry, composition, or property **a**

Advanced Search Syntax

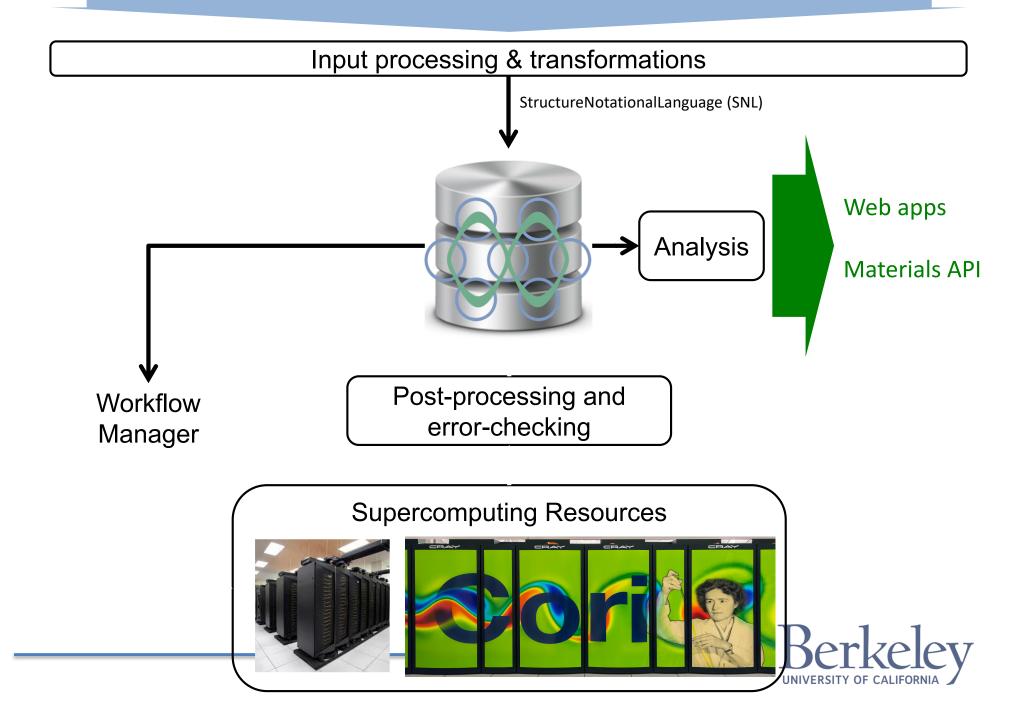
**Explore Materials** 

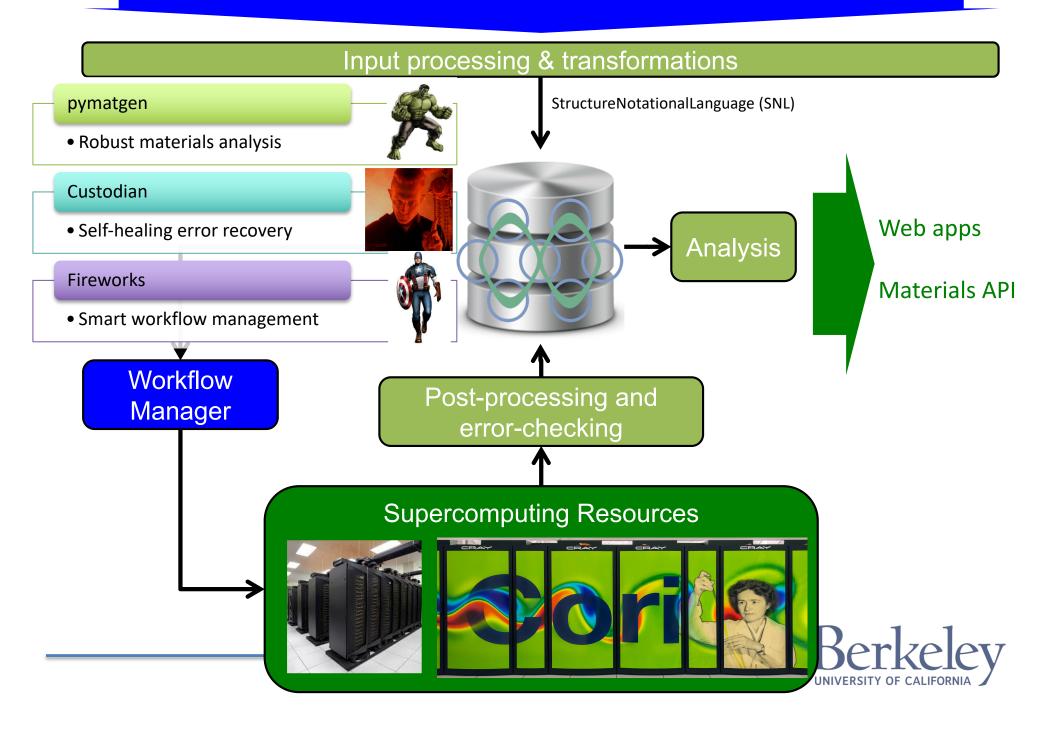
<sup>1</sup> H	 Q_t	y Formula	✓ Al2O3			×	Explanation of sin	nilarity distance measu	re: Documentation	L		
3 4	0	1 2 3	4 5	6 7 8	9 5 6							
Li E 11 12 Na M					B 13 1 Al	MATERIAL				elements 🔺 distance 🔺		
19 20	21 22		25 26 2 Mn Fe	7 28 29 Co Ni Cu	30 31	Al <sub>2</sub> O <sub>3</sub>	Piezoelectricity					
37 38	39 40	41 42	43 44 4	5 46 47			Reference for tensor and propertie	es: D Methodology				
55 56	57-71 72		75 76 7		Cd In 80 81	HM:P 1 a=4.805Å b=4.805Å c=13.116Å a=90.000° β=90.000° $\gamma$ =120.000°						
	Ba La-Lu H	If Ta W	Re Os	Ir Pt Au	Ha Tl		Piezoelectric T	ensor e <sub>ij</sub> (C/m²)	Piezoelectric Modulus Ile <sub>ij</sub> II <sub>max</sub>	Crystallographic		
							0.00000 0.00000 0.00000	0.00000 -0.15142 0.00000	0.64163 C/m <sup>2</sup>	Direction v <sub>max</sub>		
Reference for ter	nsor and pr	operties:					0.00000 0.00000 0.00000	0.57103 0.00000 0.00000		0.00000		
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F		-	s of State			l	L			1.00000		
452 150	) 10	Reference:				Calcul	lation Summary					
150 452	2 10			uations of S		Elasticity						
107 107	<b>4</b> 5	-6.5	Click and drag to zoom			Methodolog	e Optimization		, ternary, quat	ternary, quaternary, etc. structures.		
20 -20	-0				<ul> <li>computed</li> <li>mie grune</li> </ul>		opumzauon	Energy Cutoff		Amplitude:		
0 0				GGA 520 eV				0.2				
0 0	Reference for	tensor and prop	perties:	Methodology		# of K-poi	ints	U Values		Vectors:		
L	Die	lectric Tensor <i>ɛ</i>	c <sub>ij</sub> ∞ (electronic co	ontribution)	Polycı — (electı	n Beeudene	otentials	Final Energy/Atom		Speed:		
<u>t</u>	19.00	0.00		0.00	19.60		V: O AI O AI O	-7.4813 eV		pause		
Substrates	0.00	19.00	)	0.00	Polyci		eV					
Reference for mil	0.00	0.00		20.78	(total)		e arrive at this value? input parameters and outputs for all	calculations	Phonon	dispersion =		
substrate orienta	_				_ 34.18	GGA Unif	form v2 GGA band structure v2 GGA o	otimize structure (2x) GGA optimize structure (2)		dispersion =		
substrate mate	г	Dielectri	c Tensor $arepsilon_{ij}$ (tota	1)	Refrac		mize structure (2x) GGA static v2					
GaSe (mp-1943	31.99	-0.00		0.00	4.43	Why the <b>bo</b>	ld text?					
LiF (mp-1138)	-0.00	31.99	) .	-0.00	Poten	t						
GaTe (mp-5428	0.00	-0.00	:	38.56	True		ON History Show BibTex Citation Down					
SiC (mp-7631)		<1 0 0>		<1 0 1>	0.021	ICSD IDs	9775 63647 63648 93096 16	3 160904 52044 92631 52025 85137 24851 5608 1060 151589 26790 75559 75560 88028 9770 9 87 30025 52024 31545 31546 31547 31548 896	771 9772			
Mg (mp-153)		<1 0 0>		<1 0 0>	0.044		89664 89665 24005 99783 3 92629 92630 75479 165594	0024 64713 30026 30027 30028 30029 30030 4 88027 160604 160605 160606 160607 600672 60	3732 L B1 B Z	Γ X Q F P1 Z L P		
Up to 50 entries d	lisplayed.	-0 0 1		-0 0 1.	0.045							
<sup>†</sup> minimal coincide	nt interface a	area.				Submitted	by Michael Kocher Anubhav Jain	Shyue Ping Ong Geoffroy Hautier	UNIVER	RSITY OF CALIFORNIA		

## The web site is the tip of the iceberg...

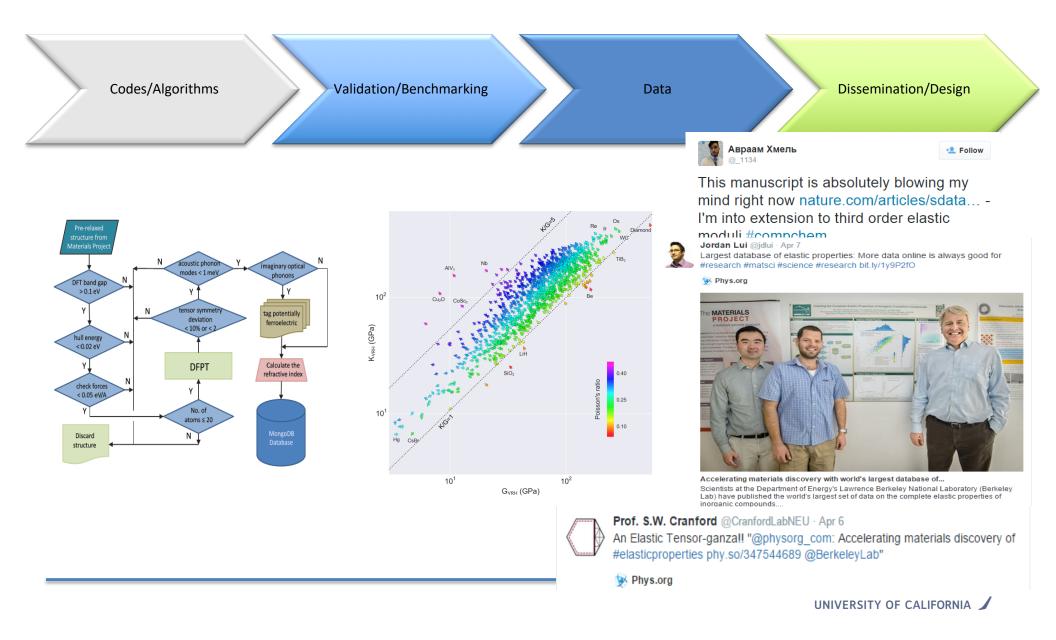




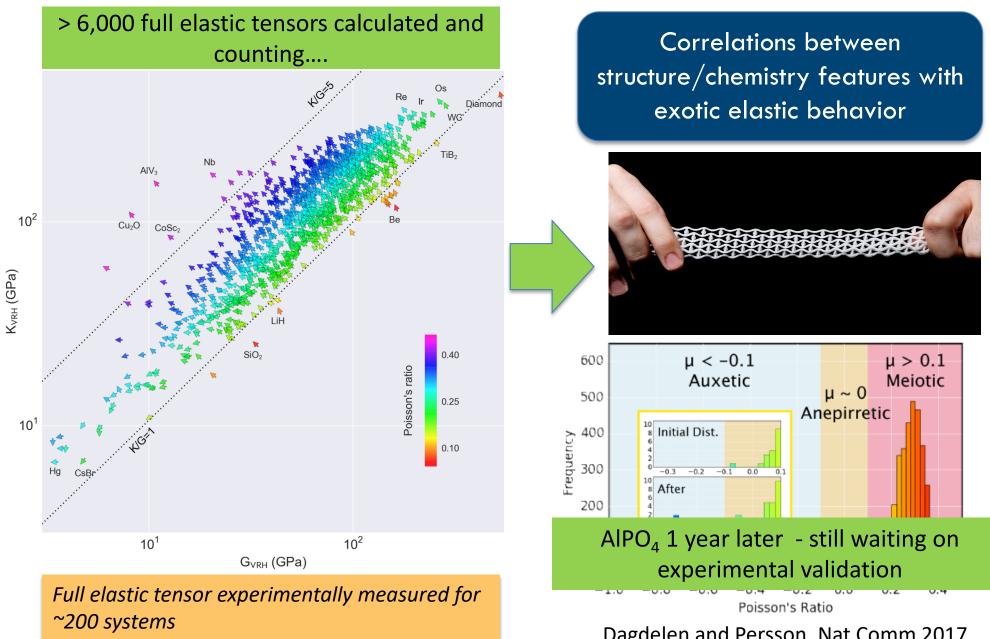




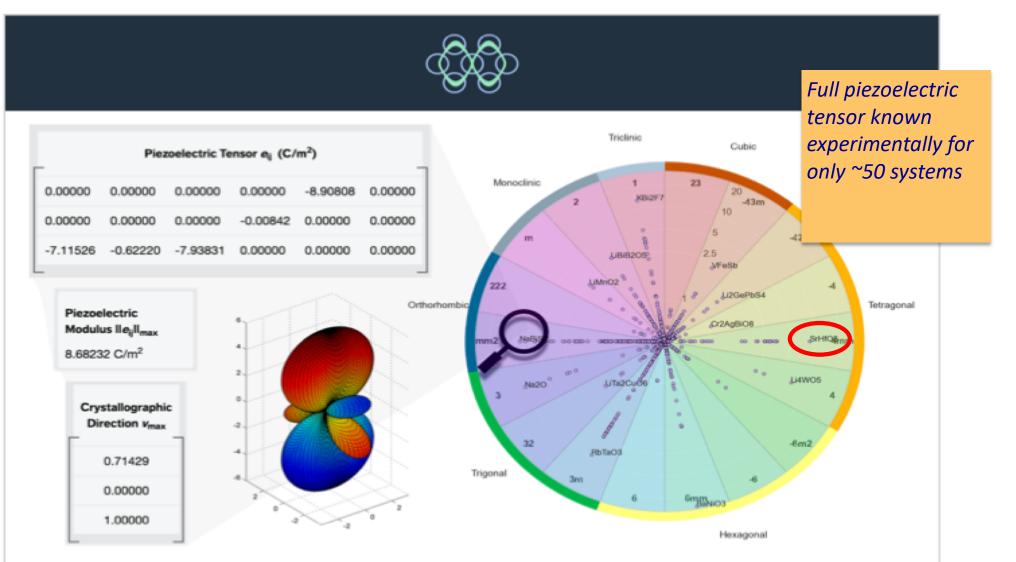
## All HT Properties Benchmarked and Automated



### Elastic Data Enables Discovery of Novel Functional Materials



Dagdelen and Persson, Nat Comm 2017



### **Piezoelectricity Data on the Materials Project**

Today, we're proud to officially launch the largest database to-date of calculated piezoelectric properties of inorganic crystalline compounds. The full piezoelectric tensor--and derived properties such as the crystallographic direction of the maximum piezoelectric response and the magnitude of the maximum response--of over 950 piezoelectric materials are now available.

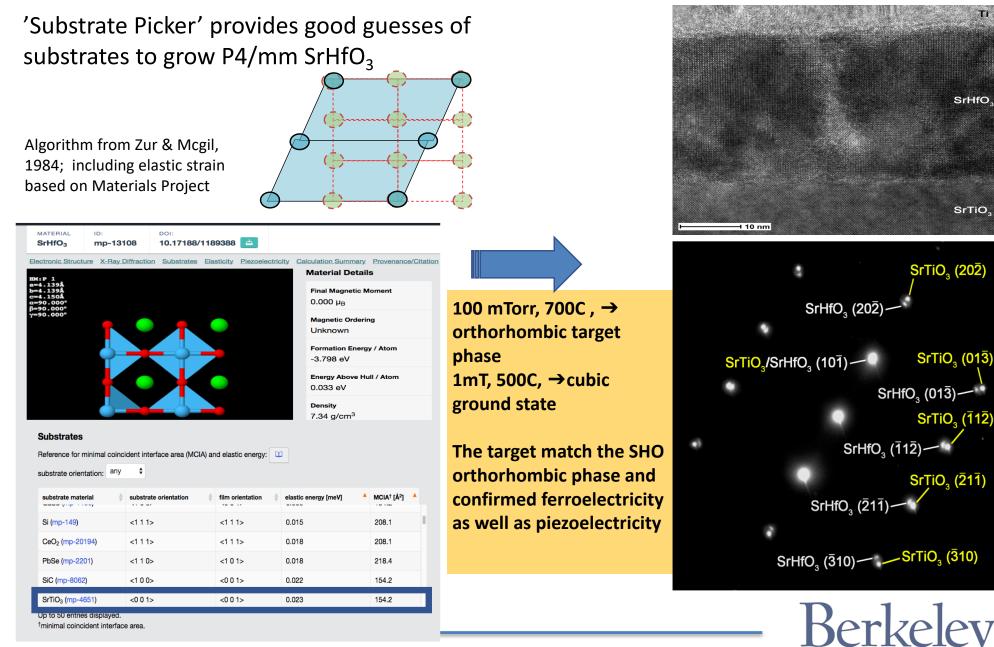
## How to target one metastable SrHfO<sub>3</sub>?

	941 Materials			E <sub>surface</sub> E <sub>polymo</sub>		epitaxial film	
Materials Id 🝦	Formula 🍦	Spacegroup 🖕	Formation Energy (eV)	E Above Hull (eV)	Band Gap (eV)		
mp-3378	SrHfO <sub>3</sub>	Pnma	-3.831	0	4.170	Groundsstate	
mp-13109	SrHfO <sub>3</sub>	Imma	-3.829	0.003	3.989	<b>Theoretical Orthorhombic</b>	
mp-4673	SrHfO <sub>3</sub>	Cmcm	-3.827	0.004	3.957	Non-Polar Orthorhombic	
mp-3721	SrHfO <sub>3</sub>	l4/mcm	-3.825	0.006	4.058	Non-Polar Tetragonal	
mp-550908	SrHfO <sub>3</sub>	P4/mbm	-3.818	0.013	3.676	Theoretical Tetragonal	
mp-13108	SrHfO <sub>3</sub>	P4mm	-3.798	0.033	3.733	Polar Tetragonal	
mp-4551	SrHfO <sub>3</sub>	Pm3m	-3.798	0.033	3.729	Cubic nonpolar Perizelev	
Iong, Dwaraknath, Garten, Ndione, Ginley and Persson, The DURING Chamistry C 120(22) 2016							

Journal of Physical Chemistry C 120(23), 2016

## **Theory-Guided Thin Film Synthesis**

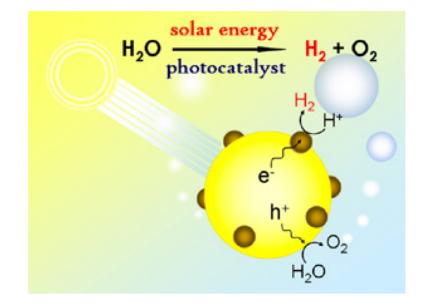
ΤÌ



Garten, Dwaraknath, ... Persson and Ginley, Submitted 2018



## Known Light Absorbers/Photocatalysts

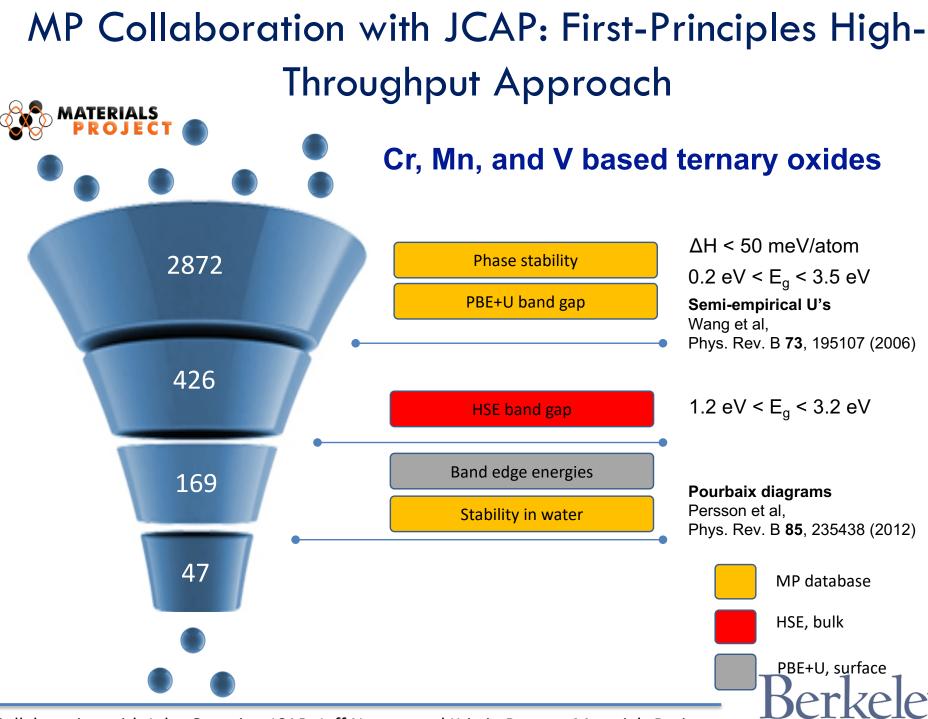


#### Criteria for a good light absorber

- (i) VBM within 0.1-0.5 eV of OER potential
- (ii) Band gap between 1 and 2 eV
- (iii) Photochemical stability at pH 13

			·	-
	(i)	(ii)	(iii)	
CdSe	Y	N	N	1
GaP	Y	Y	N	]
Si	Y	N	N	]
MoS <sub>2</sub>	Y	Y	N	1
ZnSe	Y	N	N	]
$Ta_3N_5$	Y	N	N	1
Cu <sub>2</sub> O	Y	Y	N	
CdS	Ν	N	N	]
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub>	N	N	Y	
BiVO <sub>4</sub>	N	N	N	
NaTaO₃	N	N	Y	
SrTiO₃	N	N	Y	
TiO <sub>2</sub>	Ν	N	Y	
WO <sub>3</sub>	Ν	N	N	





Collaboration with John Gregoire, JCAP; Jeff Neaton and Kristin Persson Materials Project

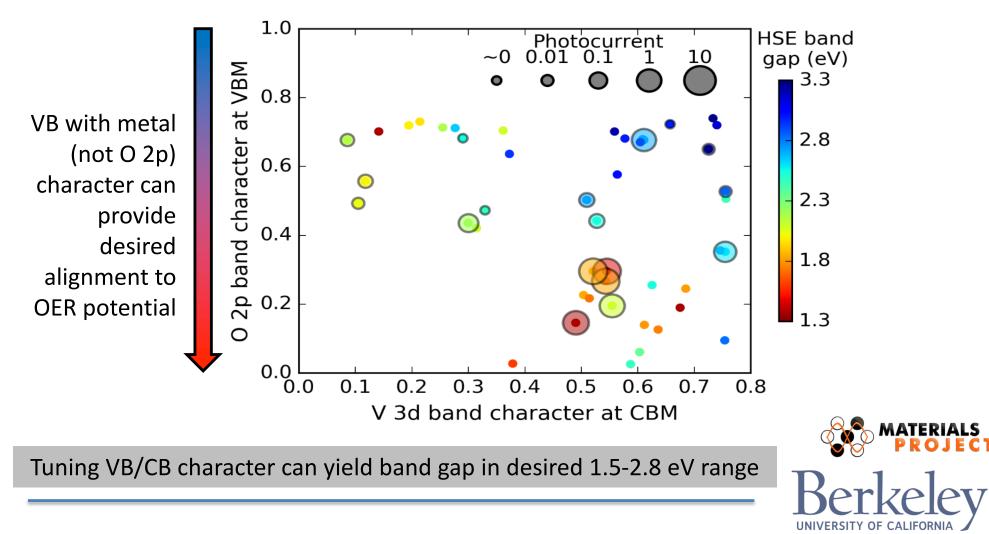
UNIVERSITY OF CALIFORNIA

## Theory Guided Design and Rapid Synthesis

• 47 phases passed computation criteria

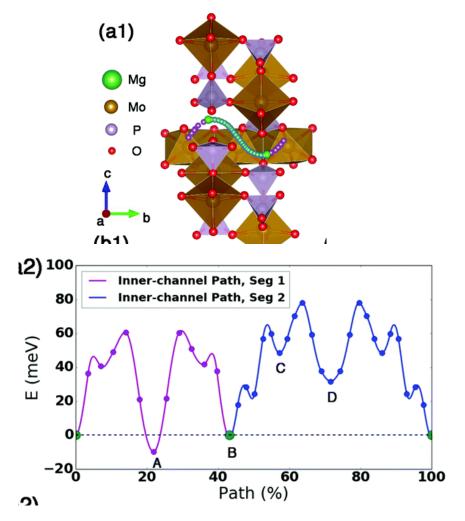
**Courtesy of John Gregoire, Caltech** 

- 17 of the 47 phases synthesized through combinatorial synthesis experiments
- 16 of these 17 phases exhibit photocurrent at OER potential



# ...however mostly: to date unrealized: Mg Cathode: $MgMo_3P_3O_{13}$

https://materialsproject.org/batteries/mvc-1200010538/



Discharged	MgMo3P3O13				
mp-id discharged	mvc-10538 (p2 <sub>1</sub> /m)				
Charged	Mo3P3O13				
mp-id charged	mvc-10615 (p2 <sub>1</sub> /m)				
<mark>max. Ehull</mark>	<mark>42 (meV/atom)</mark>				
Parent compound	CaFe3(PO4)3O				
Gravi. Capacity	87 mAh/g				
Vol. Capacity	330 mAh/cm^3				
Average Voltage	1.90V				
Diffusion Barrier	< 100meV				

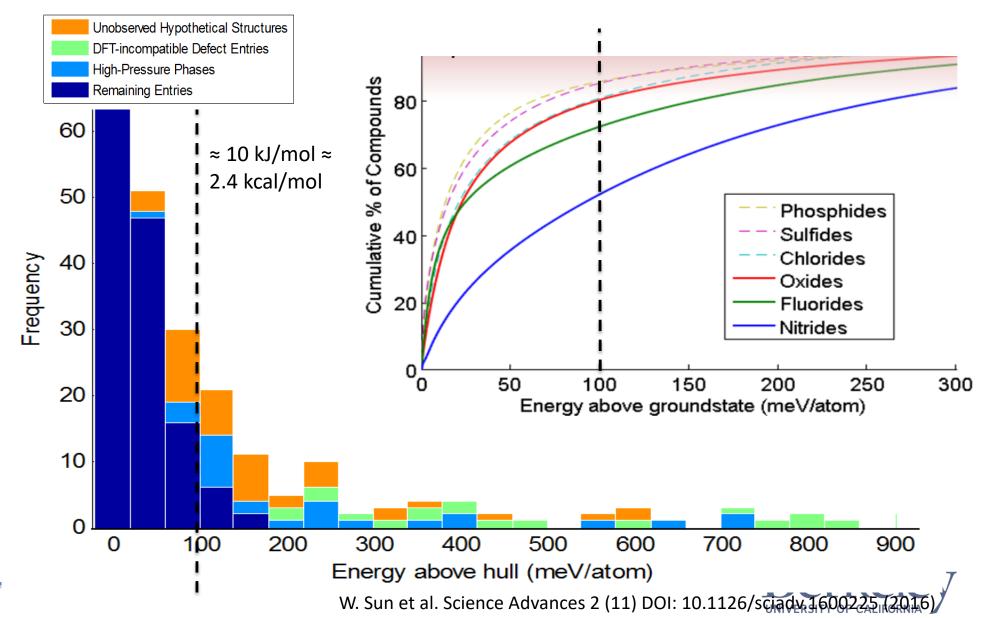


Chem. Commun., 2017, 53, 7998-8001 (Ceder&Persson group)

## How to Accelerate Synthesis of Novel Materials for Energy Applications?

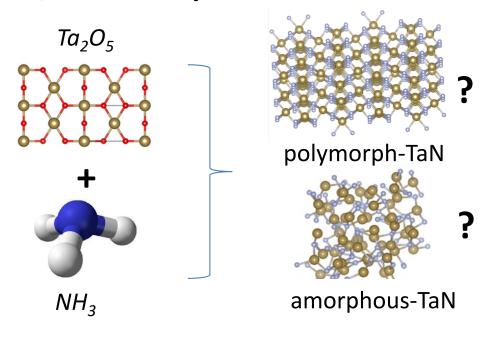


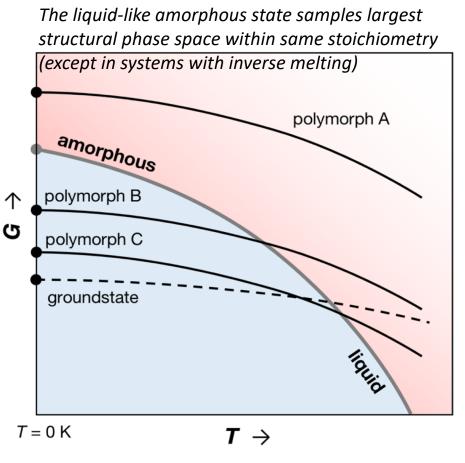
# Statistics on the ICSD covering most known inorganic materials



#### Hypothesis: the amorphous phase is always kinetically possible to form

...  $Ta_2O_5$  nanosized powder was placed in a quartz tube furnace and subjected to nitridation in flowing NH<sub>3</sub> gas at a flow rate o. 1 L/min at 8001C for 6 h. J. Am. Ceram. Soc., 88 [12] 3519–3521 (2005)



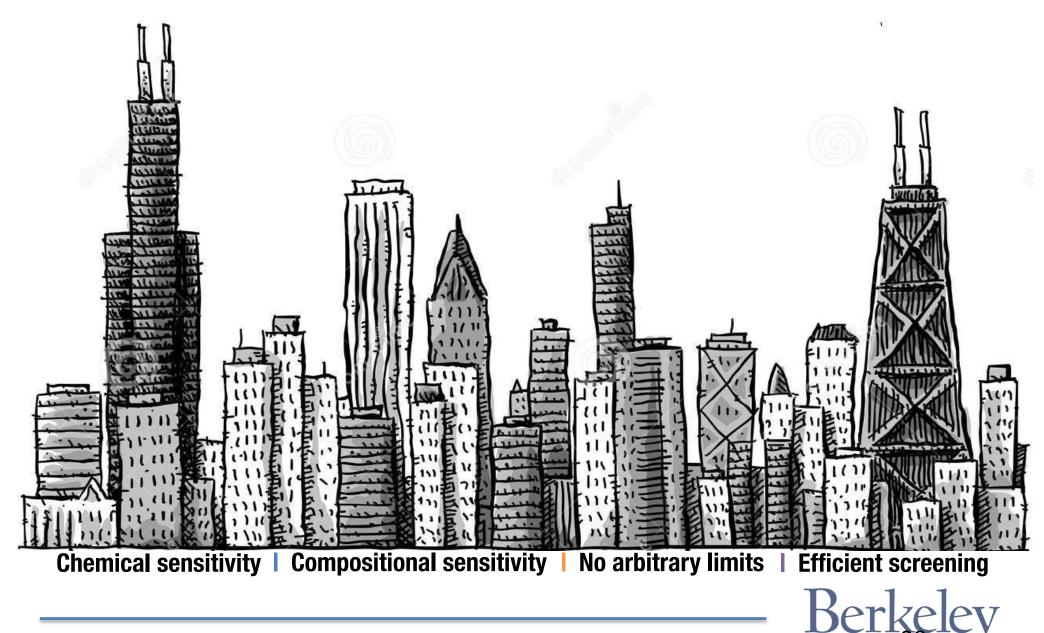


Aykol, Dwaraknath, Sun and Persson, Sci Adv 2018

Hence, an energy limit of polymorph synthesizability at any T is set by the enthalpy of the analogous amorphous state



## The synthesizability "skyline"



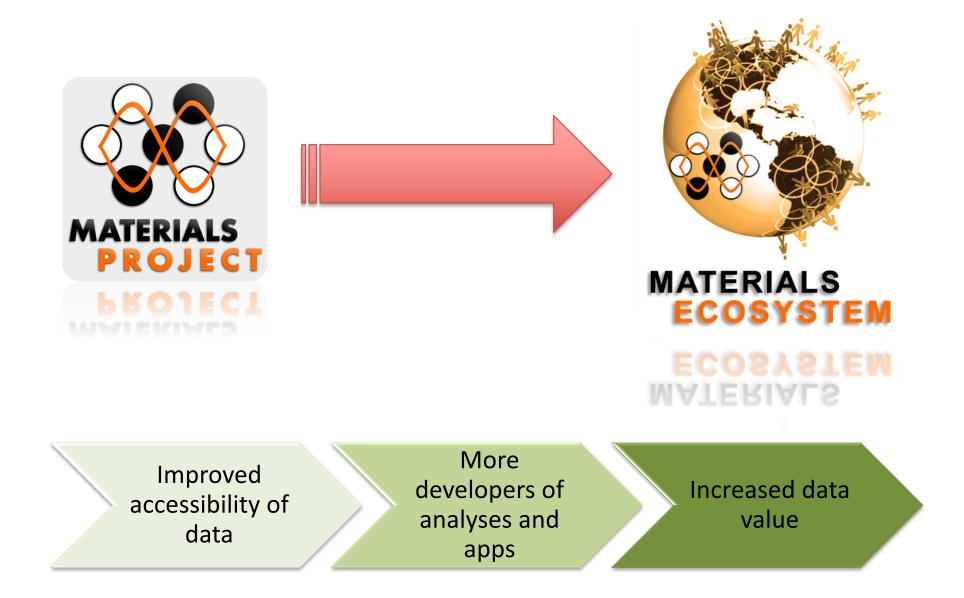
UNIVERSI

Aykol, Dwaraknath, Sun and Persson, Sci Adv 2018

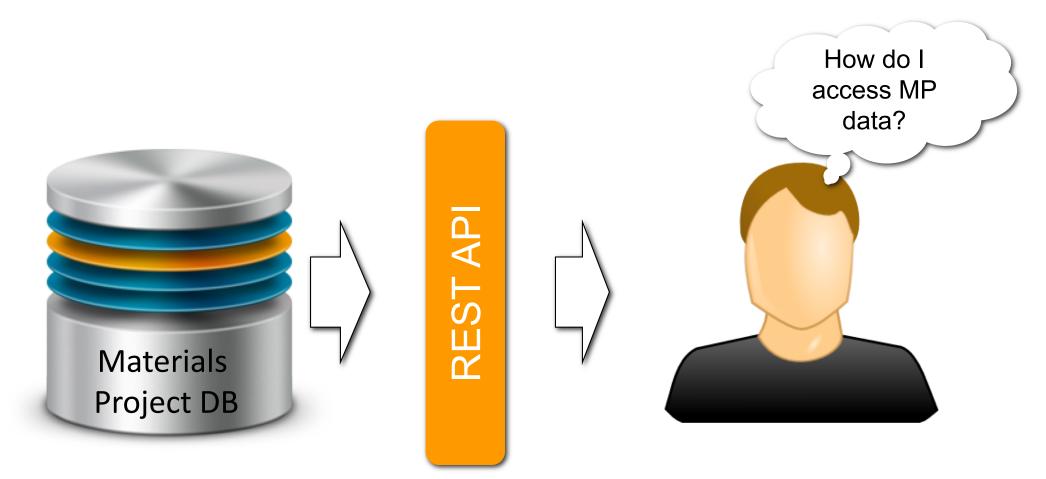


# **Build it and They will Come ?**





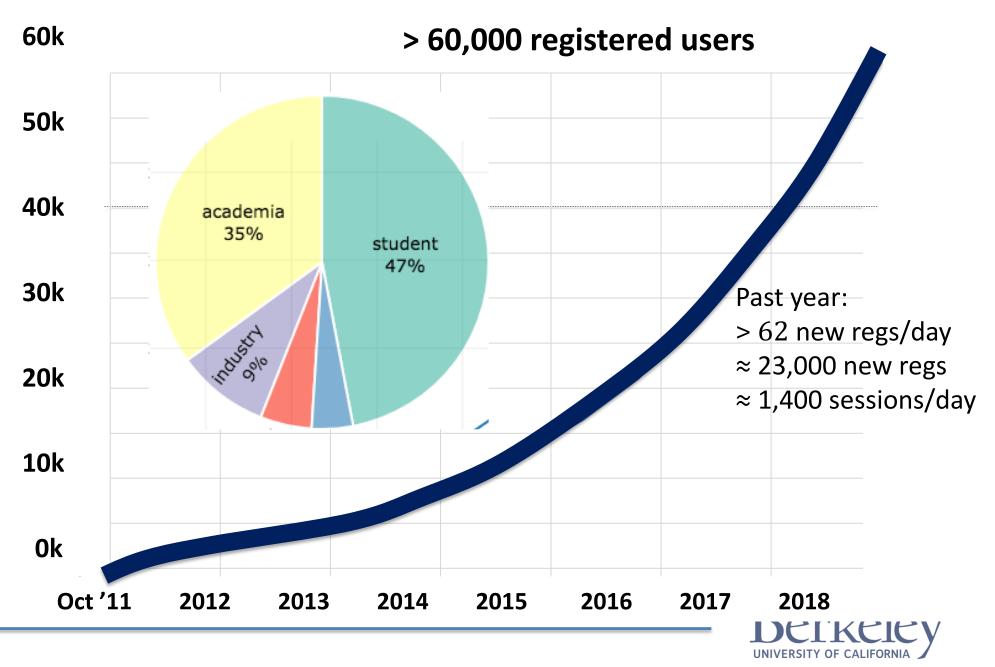
## Materials Project REST API



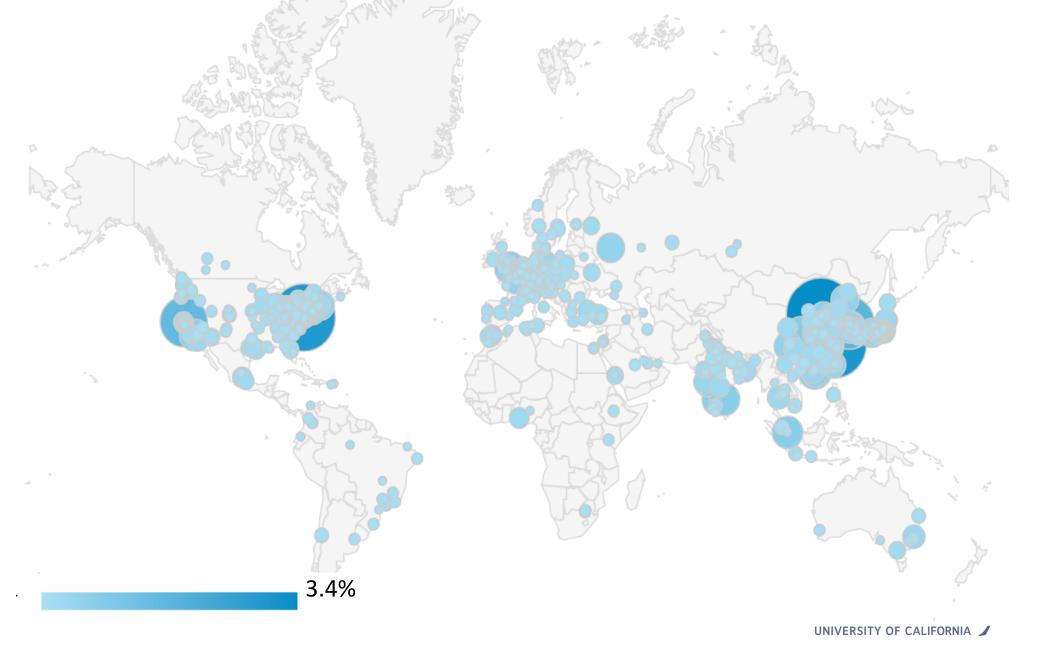
## > 160M requests served in last year



## **Rapidly Increasing Users**



## Materials Project is a Global Resource



- *First I would like to thank you a lot for this project, it is incredibly useful. I work in the field of materials design of multiphase and anisotropic materials.* (US Student)
  - *I have registered with the Materials Project and expect to try the software. Do, or can, users contribute to the library(s)?* (US Professor)
- In this framework, I would like to ask if it would be possible to organize a short training course on materials modelling using the Materials Project capabilities. (Professor in Greece)
  - Thank you very much and thank you for offering this fantastic data base! (US Student)
- I'm currently writing a website to host a database of EELS spectra (the redevelopment of EELS database if you're familiar with it). The site is similar to the Materials Project Explorer in a number of ways browsing by formula and so on it would be great if we could link out to you guys from spectra pages if possible. (UK scientist)
- I noticed the change already. Very fast response, you guys are awesome! (US student)

## Thanks to the community and for your attention ! **Office of Energy** Science

• I am a brazilian research in materials science. Firt of all I would like to congratulation for the app's they are very usefull. (Brazilian student)

• I am enjoying materialsproject.org a lot these days - it is wonderful to be able to do research without doing a single calculation. (US researcher)